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### DATA EVALUATION RECORD

#### STUDY 3

SHAUGHNESSY NO. 128994

MON-15151

Sec. 162-1

## FORMULATION-00-ACTIVE INGREDIENT

MRID NO. 40638629

Pantano, L.K. and Y.L. Sing. 1988. The aerobic soil metabolism of MON-7200/15100, 3,5-pyridine-dicarbothioic acid, 2-(difluoromethyl)-4-(2-methyl-propyl)-6-(trifluoromethyl)-,S,S-dimethyl ester. Laboratory Project No. MSL-7781. R.D. No. 866. Prepared and submitted by Monsanto Agricultural Company, St. Louis, MO.

DIRECT REVIEW TIME = 12

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CONCLUSIONS:

Metabolism - Aerobic Soil

This study is acceptable and fulfills EPA Data Requirements for Registering Pesticides by providing information on the metabolism of ring-labeled  $[^{13}\text{C}/^{14}\text{C}]\text{MON-}7200/15100$  in aerobic sandy loam, silt loam, loam, and clay soils.

#### SUMMARY OF DATA BY REVIEWER:

[13C/14C]MON-7200/15100 (labeled in the 4 position of the pyridine ring; radiochemical purity >99.0%), at ~1 ppm (50 ug/50 g dry soil), degraded with registrant-calculated half-lives of 336 days in Dupo silt loam soil, 418 days in Sarpy sandy loam soil, 490 days in Sharkey clay soil, 904 days in "volcanic ash upland" loam soil, and 946 days in "volcanic ash paddy" loam soil. The treated soils were incubated in the dark at 25°C

and 75% of 0.33 atmosphere moisture for 6 months (sandy loam, clay, and loam soils) or 12 months (silt loam soil only) posttreatment. Nonvolatile degradates, each comprising  $\leq 3.3\%$  of the applied in the five soils at intervals up to 6 months posttreatment, were . . .

2-(difluoromethyl)-4-(2-methylpropyl)-6-(trifluoromethyl)-3,5pyridinedicarboxylic acid (diacid; IV);

6-(difluoromethyl)-4-(2-methylpropyl)-5-[(methylthio)carbonyl]-2-(trifluoromethyl)-3-pyridinecarboxylic acid (reverse acid; III); and

2-(difluoromethyl)-4-(2-methylpropyl)-5-[(methylthio)carbonyl]-6-(trifluoromethyl)-3-pyridinecarboxylic acid (normal acid; II).

After 6 months of incubation, between 4.9 and 16.0% of the applied had been volatilized as undegraded MON-15151 and 0.06-0.71% had been volatilized as  $^{14}\text{CO}_2$ ; unextractable [ $^{14}\text{C}$ ] residues ranged from 3.99 to 11.10% of the applied. In the Dupo silt loam soil, the maximum concentration of the nonvolatile degradates occurred at 9 months posttreatment; the diacid accumulated to a maximum concentration of 6.9% of the applied, the reverse acid accumulated to 3.0% of the applied, and the normal acid to 5.9% of the applied. In the Dupo silt loam soil at 12 months posttreatment,  $\approx 19.6\%$  had been volatilized as MON-15151, 0.43% was  $^{14}\text{CO}_2$ , and 7.08% was unextractable. The material balances in all soils throughout the study ranged from 92.49 to 102.06% of the applied.

## DISCUSSION:

- 1. Although it was stated that additional flasks of soil were autoclaved and treated with  $[^{13}\text{C}/^{14}\text{C}]\text{MON-7200/15100}$  to serve as sterile controls, no data were provided.
- 2. The volcanic ash soils are not typical of soils in the United States. They were included in this study because MON-15151 is to be registered in Japan.

MATERIALS AND METHODS

## MATERIALS AND METHODS:

[ $^{13}\text{C}/^{14}\text{C}]$ MON-7200/15100 (labeled in the 4 position of the pyridine ring; radiochemical purity >99.0%, specific activity 18.32-26.21 mCi/mmol, Monsanto Chemical Company) was applied at  $\approx 1$  ppm (50  $\mu$ g/50 g dry soil) to flasks containing sieved (2 mm) moist (75% of 0.33 atmosphere) Sarpy sandy loam, Dupo silt loam, Sharkey clay, volcanic ash paddy loam, or volcanic ash upland loam soil (Table 5). After treatment, the soil was "lightly shaken" and the flasks were sealed with an aerobic trapping tower that was constructed of two foam plugs (in the neck of the flask), a layer of Drierite, a foam plug, Ascarite II (NaCH fused to a silicate carrier), a foam plug, Ascarite II, a foam plug, Drierite, and a foam plug (at the top of the column). All flasks of soil were incubated in the dark at 25°C; the soils were remoistened once or twice each week. The trapping towers were replaced weekly. Duplicate flasks were removed for analysis at intervals up to 6 months (Sarpy sandy loam, Sharkey clay, volcanic ash paddy loam, and volcanic ash upland loam soils) or 12 months (Dupo silt loam soil) posttreatment.

Soil samples were transferred to centrifuge tubes. The sample flasks were rinsed with acetonitrile:water (50:50), and the rinses were transferred to the appropriate centrifuge tubes. The soils were extracted by shaking for 20 minutes on a wrist action shaker, then centrifuged. The acetonitrile:water extracts were decanted and fresh acetonitrile:water was added to the soil. The soils were extracted with acetonitrile:water a total of five times; the extracts were combined and an aliquot was analyzed by ISC. Additional aliquots were analyzed by HPIC with radioactivity detection, and [14C]compounds were tentatively identified by comparison to reference standards of MON-7200/15100, Normal acid, Reverse acid, and diacid. The extracted soils were analyzed for unextractable radioactivity by ISC following combustion.

In order to confirm the identity of [\$^{14}\$C]\$compounds, the acetonitrile:water extract of the 6-month Dupo silt loam soil sample was made basic with sodium bicarbonate and partitioned three times with hexane. The aqueous fraction was acidified and partitioned three times with ether. The hexane, ether, and aqueous fractions were analyzed by LSC; the hexane and aqueous fractions were analyzed by HPIC with radioactivity detection. [\$^{14}\$C]\$compounds isolated by HPIC were identified by HPIC/MS (negative ion mode).

The foam plugs from the trapping towers were analyzed by ISC. Foam plugs from the Dupo silt loam (week 28) and volcanic ash upland loam (week 22) soil were extracted with ether; aliquots of the ether extracts and the extracted foam plugs were analyzed by ISC. Additional aliquots of the ether extracts were concentrated, dissolved in acetonitrile:water (50:50), and analyzed by HPIC and ISC. The Ascarite samples were mixed with distilled water, then sulfuric acid was added dropwise to the solution. Escaping  $\mathbf{CO}_2$  was trapped in a phenethylamine solution; the phenethylamine solution was analyzed by ISC.

# Dithiopyr Science Review

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